

# Synthesis and transformations of sulfur-substituted indolizidines and quinolizidines

Shang-Shing P. Chou\* and Chung-Wen Ho

Department of Chemistry, Fu Jen Catholic University, Taipei, Taiwan 242, ROC

Received 1 September 2005; revised 29 September 2005; accepted 30 September 2005

Available online 14 October 2005

**Abstract**—3-Sulfolenes **1a–b** underwent [4+2] cycloaddition reactions with *p*-toluenesulfonyl isocyanate to give tetrahydropyridinones **3a–b**. Through *N*-detosylation of **3a–b** and subsequent intramolecular cyclization, indolizidine **5a** and quinolizidine **5b** were synthesized. Useful functional group transformations of compounds **5a–b** were also investigated.

© 2005 Elsevier Ltd. All rights reserved.

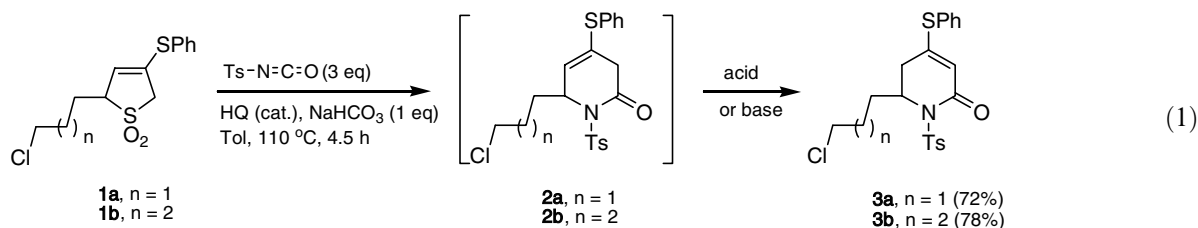
The piperidine ring is among the most abundant molecular fragments in both natural and synthetic compounds with various biological activities.<sup>1</sup> The aza-Diels–Alder reaction is one of the most versatile routes to substituted piperidines.<sup>2–7</sup> In general, the use of strongly electron-deficient imines is a prerequisite. We have also used this method to synthesize some thio-substituted piperidine derivatives.<sup>8–10</sup>

Although arylsulfonyl isocyanates have an electron-deficient C=N moiety, their aza-Diels–Alder reactions with dienes were rarely reported<sup>11–13</sup> because the [2+2] cycloaddition or electrophilic substitution predominates.<sup>14–16</sup> We have recently reported the first aza-Diels–Alder reactions of thio-substituted dienes with arylsulfonyl isocyanates to give the cyclized products with complete control of chemo- and regioselectivity.<sup>17,18</sup> We have also used this methodology to synthesize many 6-substituted 2-piperidinones.<sup>19</sup> Two of these compounds were further converted to the indolizidine and quinolizidine

structures, which are important framework of many natural products.<sup>20–26</sup> We now report a new method for the synthesis of sulfur-substituted indolizidines and quinolizidines, and some of their functional group transformations.

Thio-substituted 3-sulfolenes **1a–b**<sup>27</sup> can undergo in situ thermal desulfonylation and [4+2] cycloaddition reaction<sup>19</sup> with *p*-toluenesulfonyl isocyanate (PTSI) to give the cyclized products **3a–b** (Eq. 1).

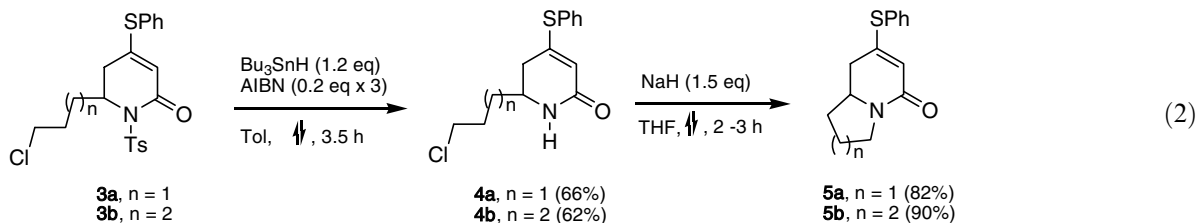
In order to synthesize the indolizidine and quinolizidine bicyclic compounds, we used Parsons' method of Bu<sub>3</sub>SnH/AIBN to cleave the *N*-tosyl group of amides **3a–b**.<sup>28</sup> Lactams **4a–b** were obtained in moderate yield (Eq. 2). It should be noted that slow addition of Bu<sub>3</sub>SnH to a dilute solution of **3a–b** was necessary in order to avoid simultaneous cleavage of the chloro group of **3a–b**. Upon treatment of lactams **4a–b** with NaH in refluxing THF indolizidine **5a** and quinolizidine **5b** were



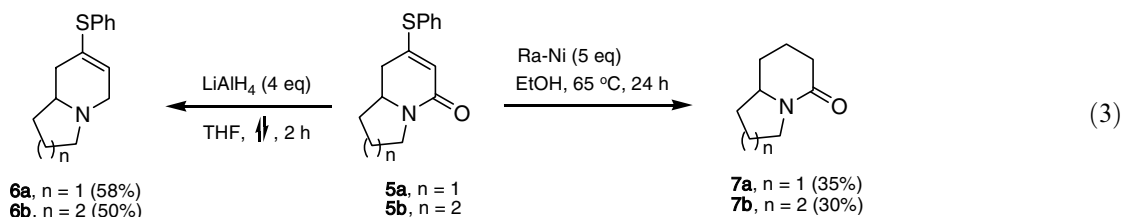
**Keywords:** Thio-substituted 3-sulfolenes; Aza-Diels–Alder reaction; Indolizidines; Quinolizidines.

\* Corresponding author. Tel.: +886 2 29052474; fax: +886 2 29023209; e-mail: chem1004@mails.fju.edu.tw

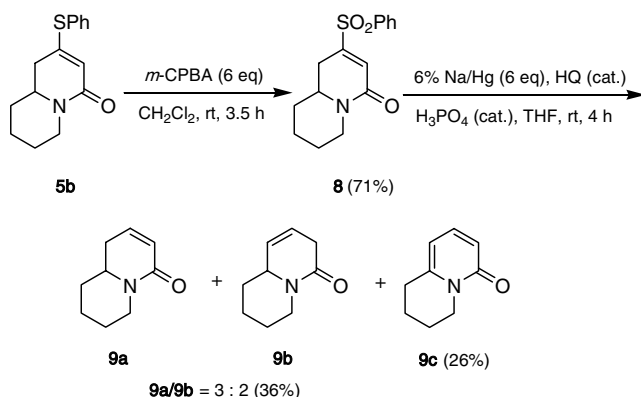
synthesized by intramolecular cyclization<sup>29–31</sup> in good yield. It is important to note that the use of potassium *t*-butoxide as the base led only to the recovered starting material or elimination of HCl to form the terminal alkenes.



With indolizidine **5a** and quinolizidine **5b** in hand, we have also carried out some functional group transformations. The lactam functional group of compounds **5a–b** was reduced by  $\text{LiAlH}_4$  to give the amines **6a–b** (Eq. 3). Treatment of compounds **5a–b** with W-2 Raney nickel both cleaved the C–S bond and reduced the C=C bond to give products **7a**<sup>32</sup> and **7b**.<sup>33</sup>



Compound **5b** was oxidized by *m*-CPBA to give sulfone **8** (Eq. 4), which upon treatment with sodium amalgam led to the sulfone-cleaved products: an inseparable mixture of **9a**<sup>34</sup> and **9b**,<sup>35</sup> together with some oxidation product **9c**<sup>36</sup> which was obtained during chromatographic purification of compounds **9a–b**. Attempted desulfurization using the standard procedure,<sup>37</sup>  $\text{Na(Hg)/MeOH/Na}_2\text{HPO}_4$ , did not proceed at all. Thus, catalytic amount of  $\text{H}_3\text{PO}_4$  in THF was used in this reaction.<sup>38</sup>



Following a reported procedure,<sup>39</sup> treatment of compounds **5a–b** with methyllithium, followed by acidification with acetic acid and reduction with sodium borohydride, gave stereospecifically the *cis* products **10a–b** in fair to good yields (Eq. 5).

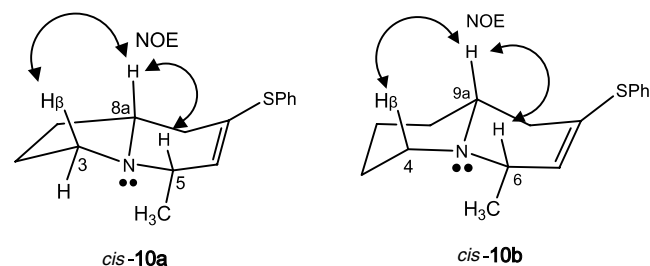
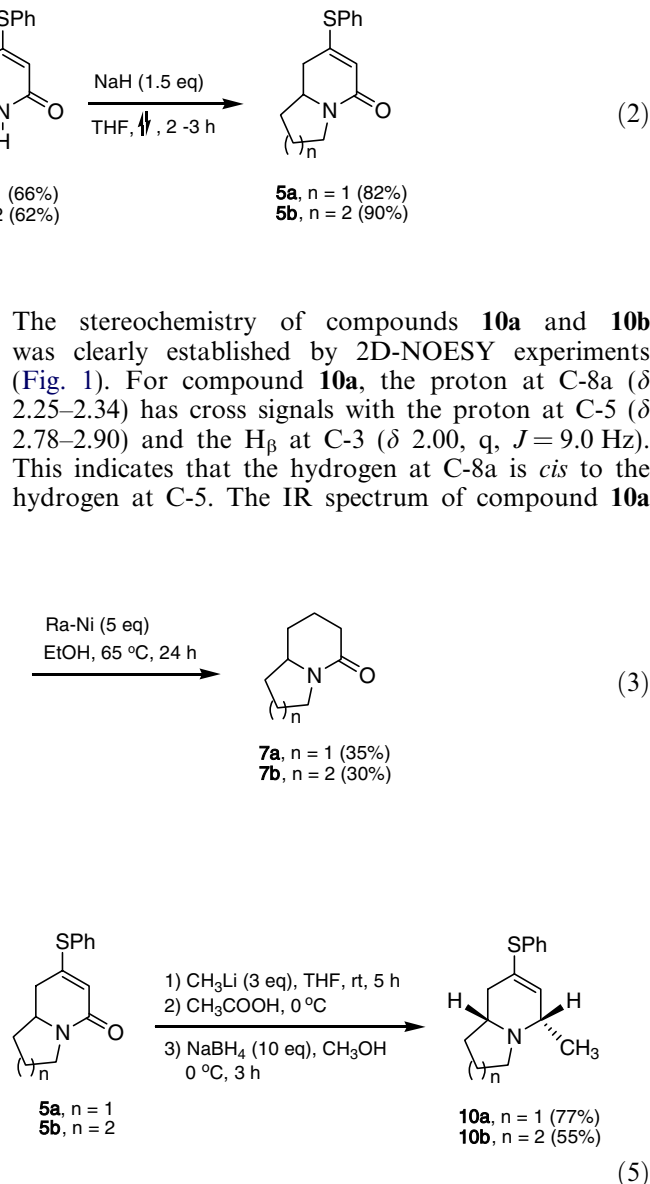
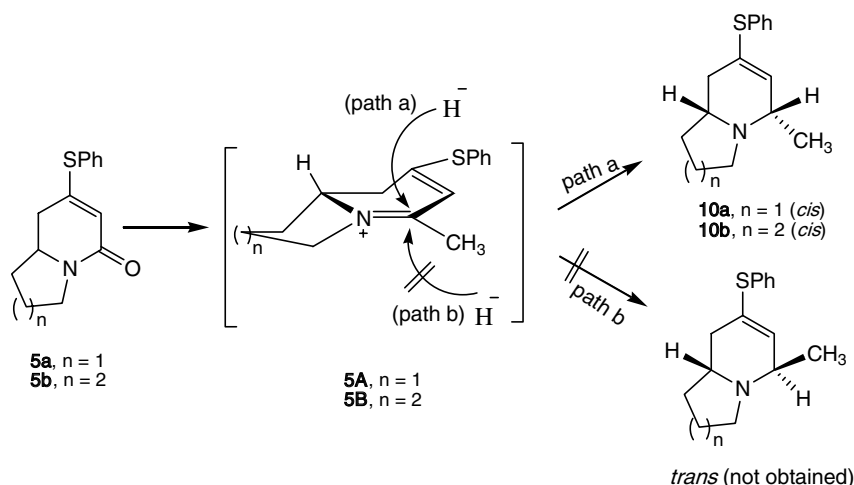


Figure 1. NOESY correlations of *cis*-**10a** and *cis*-**10b**.



**Scheme 1.** Preferred pathway for the diastereoselective formation of *cis*-**10a–b** from **5a–b**.

shows a Bohlmann band at  $2783\text{ cm}^{-1}$ , indicating the presence of one or more  $\alpha$ -hydrogens oriented *trans*-biaxially to the nitrogen lone pair.<sup>40,41</sup> Likewise, for compound **10b**, the proton at C-9a ( $\delta$  2.12–2.28) has cross signals with the proton at C-6 ( $\delta$  2.73–2.86) and the  $H_\beta$  at C-4 ( $\delta$  1.79, t,  $J = 11.3\text{ Hz}$ ), indicating the *cis* relationship of the hydrogens at C-9a and C-6. The IR spectrum of **10b** also shows a Bohlmann band at  $2788\text{ cm}^{-1}$ .

To explain the stereospecific formation of *cis*-**10a** and *cis*-**10b**, we postulated that the iminium ions **5A** and **5B** were first formed as the intermediate. The hydride ion would then prefer to attack from the axial direction due to the stereoelectronic effect (Scheme 1).<sup>42,43</sup> In addition, approach of the hydride from the  $\beta$ -face leading to a chair conformation is favored over that from the  $\alpha$ -face, because the latter will lead to a less stable boat conformation.

In summary, we have synthesized 6-substituted 2-piperidinones **3a–b** from the aza-Diels–Alder reactions of 3-sulfolenes **1a–b** with *p*-toluenesulfonyl isocyanate, and have successfully converted them to indolizidine **5a** and quinolizidine **5b**. We have also carried out some useful functional transformations of compounds **5a** and **5b**.

### Acknowledgements

Financial support of this work by the National Science Council of the Republic of China is gratefully acknowledged (NSC 93-2113-M-030-004).

### References and notes

- Rubiralta, M.; Giralt, E.; Diez, E. *Piperidine, Structure, Preparation, Reactivity and Synthetic Applications of Piperidine and its Derivatives*; Elsevier: Amsterdam, 1991.
- Weinreb, S. M. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 5, pp 401–449.
- Boger, D. L.; Weinreb, S. M. *Hetero Diels–Alder Methodology in Organic Synthesis*; Academic Press: Orlando, 1987.
- Bailey, P. D.; Millwood, P. A.; Smith, P. D. *Chem. Commun.* **1998**, 633–640.
- Laschat, S.; Dickner, T. *Synthesis* **2000**, 1781–1813.
- Jorgensen, K. A. *Angew. Chem., Int. Ed.* **2000**, 39, 3558–3588.
- Buonora, P.; Olsen, J.-C.; Oh, T. *Tetrahedron* **2001**, 57, 6099–6138.
- Chou, S. S. P.; Hung, C. C. *Synth. Commun.* **2001**, 31, 1097–1104.
- Chou, S. S. P.; Hung, C. C. *Synth. Commun.* **2002**, 32, 3119–3126.
- Chou, S. S. P.; Chen, K. W. *Synth. Commun.* **2004**, 34, 4573–4582.
- Gompper, R.; Heinemann, U. *Angew. Chem., Int. Ed. Engl.* **1980**, 19, 216–217.
- Barluenga, J.; Aznar, F.; Fernandez, M. *Tetrahedron Lett.* **1995**, 36, 6551–6554.
- For the reaction of an aza-diene with the C=N bond of tosyl isocyanate, see: Saito, T.; Kimura, H.; Soda, T.; Karakasa, T. *Chem. Commun.* **1997**, 1013–1014.
- Ulrich, H. *Chem. Rev.* **1965**, 65, 369–376.
- Arbuzov, B. A.; Zobova, N. N. *Synthesis* **1974**, 461–476.
- Takaki, K.; Okamura, A.; Ohshiro, Y.; Agawa, T. *J. Org. Chem.* **1978**, 43, 402–405.
- Chou, S. S. P.; Hung, C. C. *Tetrahedron Lett.* **2000**, 41, 8323–8326.
- Chou, S. S. P.; Hung, C. C. *Synthesis* **2001**, 2450–2462.
- Chou, S. S. P.; Chiu, H. C.; Hung, C. C. *Tetrahedron Lett.* **2003**, 44, 4653–4655.
- Daly, J. W.; Spande, T. F. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W., Ed.; Wiley: New York, 1986; Vol. 3, Chapter 1.
- For a recent review, see: Michael, J. P. *Nat. Prod. Rep.* **2004**, 21, 625–649. For some recent examples on the synthesis of indolizidines and quinolizidines, see Refs. 22–26.
- Harris, J. M.; Padwa, A. *J. Org. Chem.* **2003**, 68, 4371–4381.
- Back, T. G.; Hamilton, M. D.; Lim, V. J. J.; Parvez, M. *J. Org. Chem.* **2005**, 70, 967–972.
- Patil, N. T.; Pahadi, N. K.; Yamamoto, Y. *Tetrahedron Lett.* **2005**, 46, 2101–2103.

25. Katoh, M.; Mizutani, H.; Honda, T. *Tetrahedron Lett.* **2005**, 46, 5161–5163.
26. Maloney, K. M.; Danheiser, R. L. *Org. Lett.* **2005**, 7, 3115–3118.
27. Chou, S. S. P.; Tsao, H. J.; Lee, C. M.; Sun, C. M. *J. Chin. Chem. Soc.* **1993**, 40, 53–57.
28. Parsons, A. F.; Pettifer, R. M. *Tetrahedron Lett.* **1996**, 37, 1667–1670.
29. Nagao, Y.; Dai, W. M.; Ochiai, M.; Tsukagoshi, S.; Fujita, E. *J. Org. Chem.* **1990**, 55, 1148–1156.
30. Pourashraf, M.; Delair, P.; Rasmussen, M. O.; Greene, A. E. *J. Org. Chem.* **2000**, 65, 6966–6972.
31. Dieter, R. K.; Watson, R. *Tetrahedron Lett.* **2002**, 43, 7725–7728.
32. Park, S. H.; Kang, H. J.; Ko, S.; Park, S.; Chang, S. *Tetrahedron: Asymmetry* **2001**, 12, 2621–2624.
33. Milligan, G. L.; Mossman, C. J.; Aube, J. *J. Am. Chem. Soc.* **1995**, 117, 10449–10459.
34. Rouden, J.; Seitz, T.; Lemoucheux, L.; Lasne, M.-C. *J. Org. Chem.* **2004**, 69, 3787–3793.
35. Nukui, S.; Sodeoka, M.; Shibasaki, M. *Tetrahedron Lett.* **1993**, 34, 4965–4968.
36. Thomas, E. W. *J. Org. Chem.* **1986**, 51, 2184–2191.
37. Trost, B. M.; Arndt, H. C.; Strege, P. E.; Verhoeven, T. R. *Tetrahedron Lett.* **1976**, 3477–3478.
38. Chou, S. S. P.; Sun, C. M. *Tetrahedron Lett.* **1990**, 31, 1035–1038.
39. Yuguchi, M.; Tokuda, M.; Orito, K. *Bull. Chem. Soc. Jpn.* **2004**, 77, 1031–1032.
40. Bohlmann, F. *Chem. Ber.* **1958**, 91, 2157–2167.
41. Sonnet, P. E.; Oliver, J. E. *J. Heterocycl. Chem.* **1975**, 12, 289–294.
42. Deslongchamps, P. *Stereoelectronic Effects in Organic Chemistry*; Pergamon: New York, 1983; p 211.
43. Stevens, R. V. *Acc. Chem. Res.* **1984**, 17, 289–296.